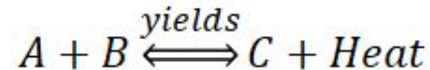


Use of a Salt Hydrate for Thermochemical Energy Storage

David Martin, Dane Hansen, Drake Graham

Background

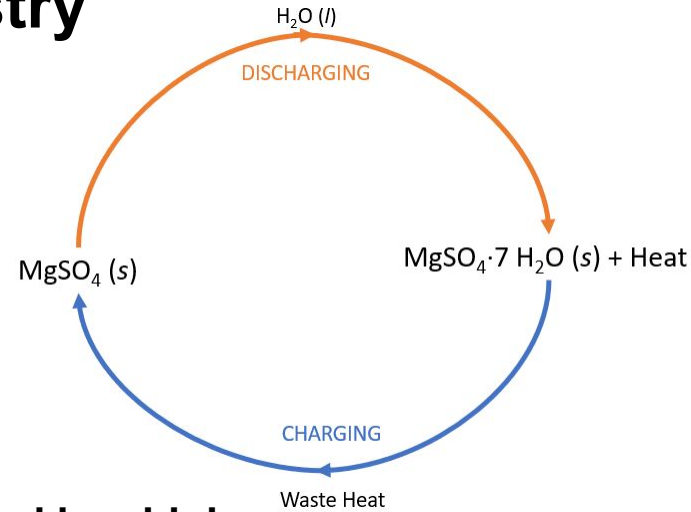
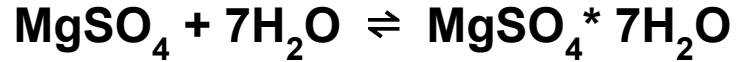
- Waste heat is produced by the Carty Power Plant in Boardman, Oregon
- This heat is normally dissipated and lost to the environment
- One way to store this heat for later use is through Thermochemical Energy Storage
- This method uses reversible chemical reactions to store thermal energy indefinitely
- Waste heat is used to drive an endothermic reaction which can be then reversed at a later time, releasing the stored energy



Current Methods

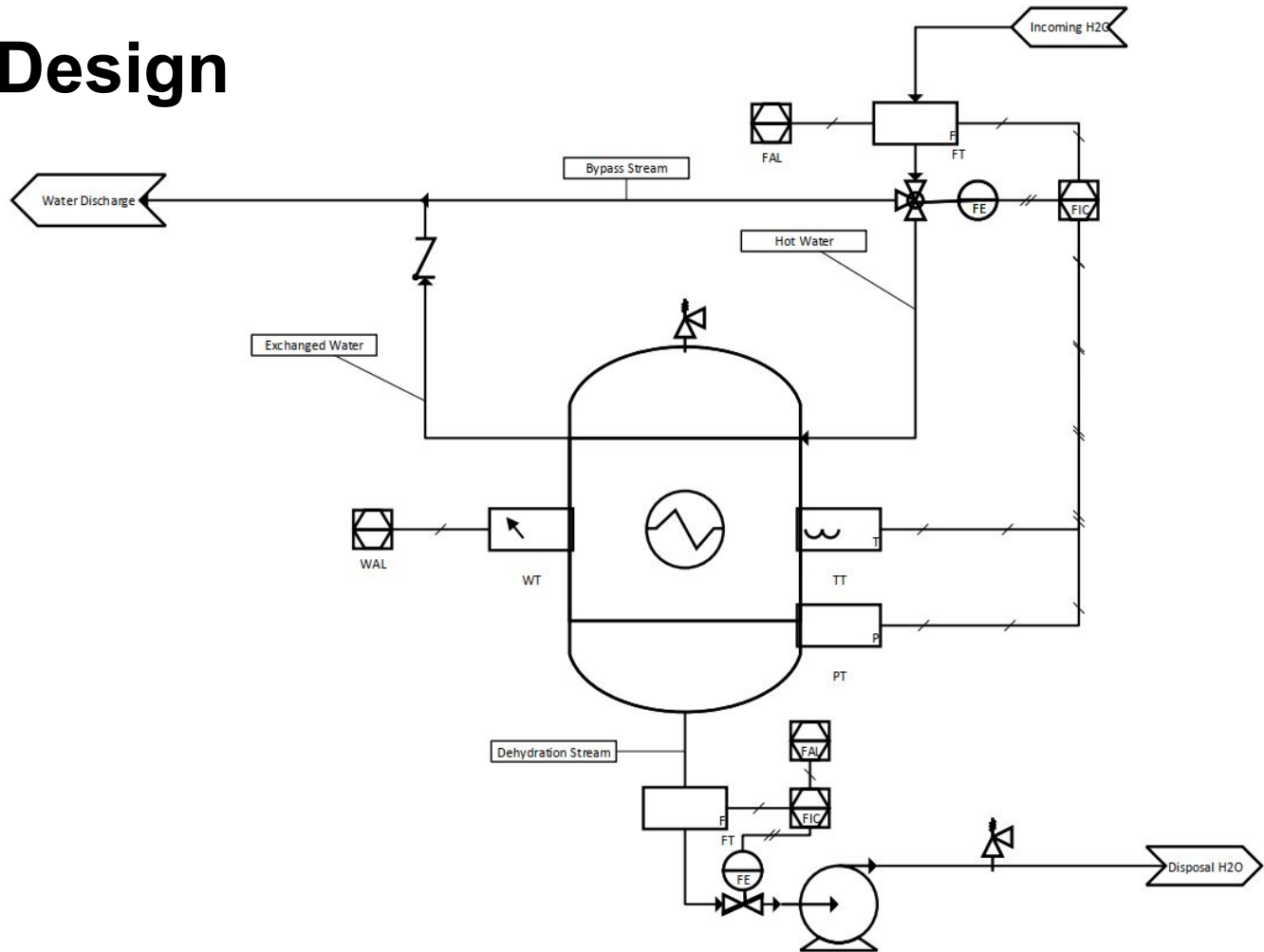
- Our design uses Magnesium Sulfate (MgSO_4), also known as Epsom Salts as the chemical basis
- To discharge our reactor, a water stream is ran through the anhydrous MgSO_4 , which causes it to transition to its heptahydrate form, releasing heat that can be then used for space heating
- To charge the reactor, the waste heat from the plant is used to heat the $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ to 150°C , for 9 days causing it to dehydrate.

TCES Reaction Chemistry



- Salt hydrate reaction
- MgSO_4 is safe, readily available, cost effective and has high energy storage density
- Charging involves dehydration reaction, discharging involves hydration reaction

Reactor Design



Operating Procedure

- Once the reactor bed has been packed and loaded the hot waste water can be diverted to the reactor
- It takes 52 minutes for the reactor to reach reaction temperature from room temperature
- For the next 9.2 days the $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ slowly absorbs heat from the waste water stream until the MgSO_4 is dehydrated
- Once dehydrated the MgSO_4 can be stored in a dry environment.
- Reactor is then cleaned and reset for next batch

Results

- Over the course of 9 days we are able to process 4m^3 of $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$
- This corresponds to 4 Gigajoules of stored energy
- Since energy generated in the 5 operational summer months is stored a total of 65.1 Gigajoules can be stored for heating in winter
- A total of 143.2 Gigajoules can be produced yearly using this reactor.

Improvements

- **At current design we can only source 7% of heating requirements**
- **To improve the efficiency of the design it would be easiest to provide a different heat exchange fluid.**
- **Reactor geometry can be further improved to squeeze out additional heat transfer.**
- **At hotter temperatures a different more efficient chemistry base can be used.**

