# PYROLYSIS OF HAZELNUT SHELLS - TEAM 1.2

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### Hazelnut Biochar Objective

- Design pyrolysis unit to produce biochar
- Constraints
  - 10,000 tonsproduced byCascade Foods
  - September to April hazelnut processing period
  - Moisture content from 20-40% to 9%



### Background

- Oregon production 65,000 tons
- Cascade Foods, LLC .
  - Located in Albany Oregon, Cascade foods is one of the largest hazelnut processers in the United States.
- Offers many products of hazelnuts including:
  - Whole Kernel
  - Roasted
  - In-shell
  - Diced
  - Organic

### Biochar

■ What is Biochar and why is it important?

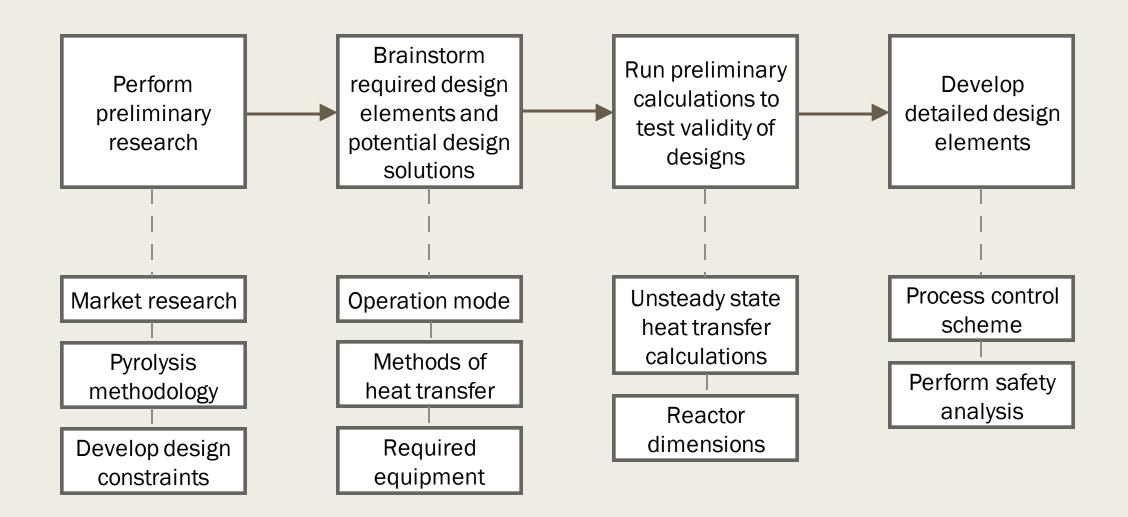


### Pyrolysis

What is pyrolysis and why is it important?



### Design Methodology



### Sample Calculations

#### Assumptions:

- Unsteady-state heat transfer, forced convection
- 1D HX in the x-direction of the shell
- No heat generation
- Constant properties during heating process
- Shell is thin enough to approximate as a shell
- Temperature bounds for heating: T = 345°C,  $T_{\infty} = 350$ °C,  $T_{o} = 20$ °C
- Temperature bounds for cooling: T = 65°C,  $T_{\infty} = 20$ °C,  $T_{o} = 345$ °C

#### Calculations: Finding pyrolysis time for shells

$$\frac{V}{A} = \frac{t}{2} \approx 0.65 \times 10^{-3} \ mm$$

$$Bi = \frac{h\left(\frac{V}{A}\right)}{k} = 0.0876$$

$$\frac{T - T_{\infty}}{T_o - T_{\infty}} = e^{-BiFo} \implies Fo = \frac{-\ln\left(\frac{T - T_{\infty}}{T_o - T_{\infty}}\right)}{Bi} = 29.65$$

$$Fo = \frac{\alpha t_{pyro}}{\left(\frac{t}{2}\right)^2} \qquad \Rightarrow t_{pyro} = \frac{Fo\left(\frac{t}{2}\right)^2}{\alpha} = 77.8 \text{ s} = 1.3 \text{ min}$$

#### Calculations: Reactor Dimensions

Assuming L/D ratio of 3, extra 20% volume for headspace

$$\frac{V_{day}}{2} = \frac{26.2 \ tons}{2} = 11884 \ \frac{kg}{batch} = 29.7 \ \frac{m^3}{batch} \rightarrow 35.6 \frac{m^3}{batch}$$

$$D = \sqrt[3]{\frac{4}{3\pi}V} = 2.47 \, m$$

$$L = 3D = 7.42 m$$

#### Calculations: Batch Time

 $t_{batch} = t_{heat} + t_{pyro} + t_{cool} + t_{transport/reset}$ 

$$t_{heat} = \frac{350^{\circ}\text{C} - 20^{\circ}\text{C}}{5^{\circ}\text{C/min}} = 66.0 \text{ min}$$

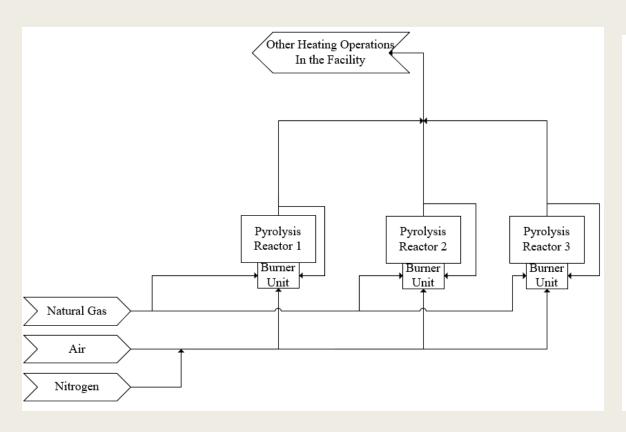
$$t_{pyro} = 78.1 s = 1.3 min$$

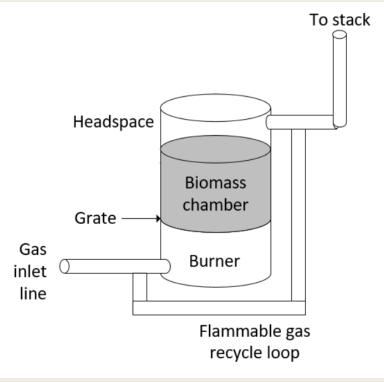
$$t_{cool} = 112.6 s = 1.9 min$$

 $t_{transport/reset} = 2 \ hours = 120 \ min$ 

#### $t_{hatch} = 190 min = 3.2 hours$

### Proposed Design





## QUESTIONS?

Come to our Engineering Virtual Expo zoom room on June 4th, 2021 from 8:30-11:30am!