

BACKGROUND HISTORY: THE DREAM OF INFINITE ENERGY

- Bhāskara II's Wheel (circa 1150 CE [1])

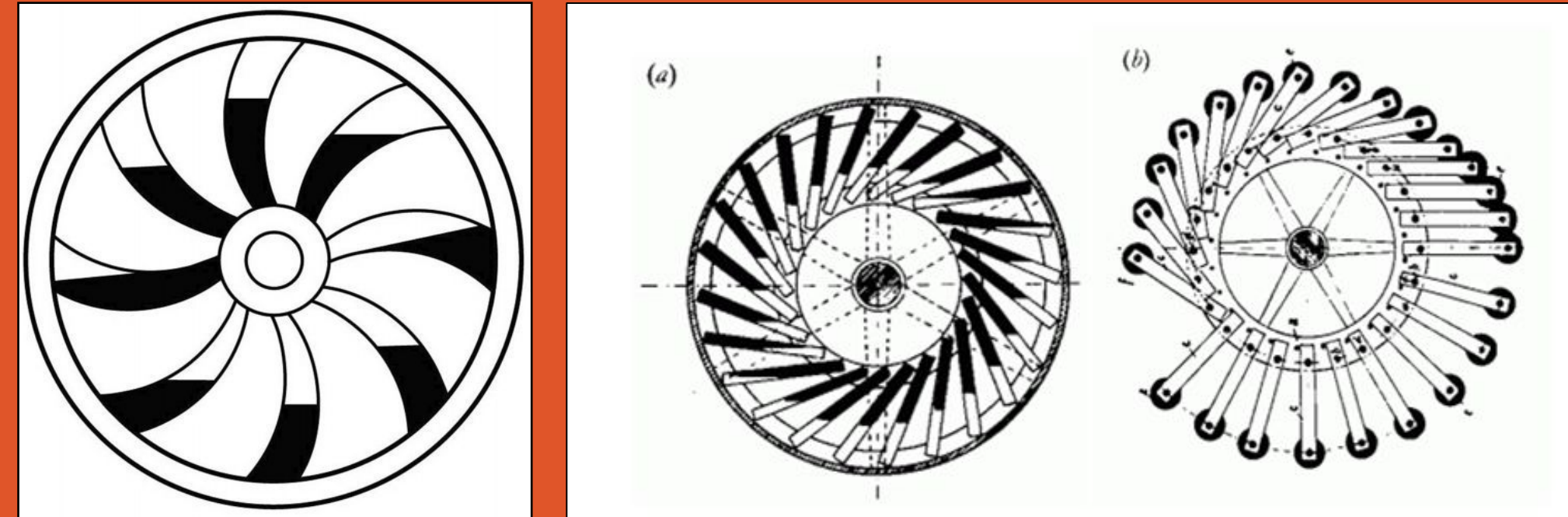


Figure 1: Diagram Depiction of Bhāskara II's Wheel Figure 2: Diagrams of Changing Centers of Mass PMMs

- Mariano di Jacopo, a.k.a. "Taccola" (circa 1400 [2])
- Leonardo Da Vinci (circa 1500 [3])

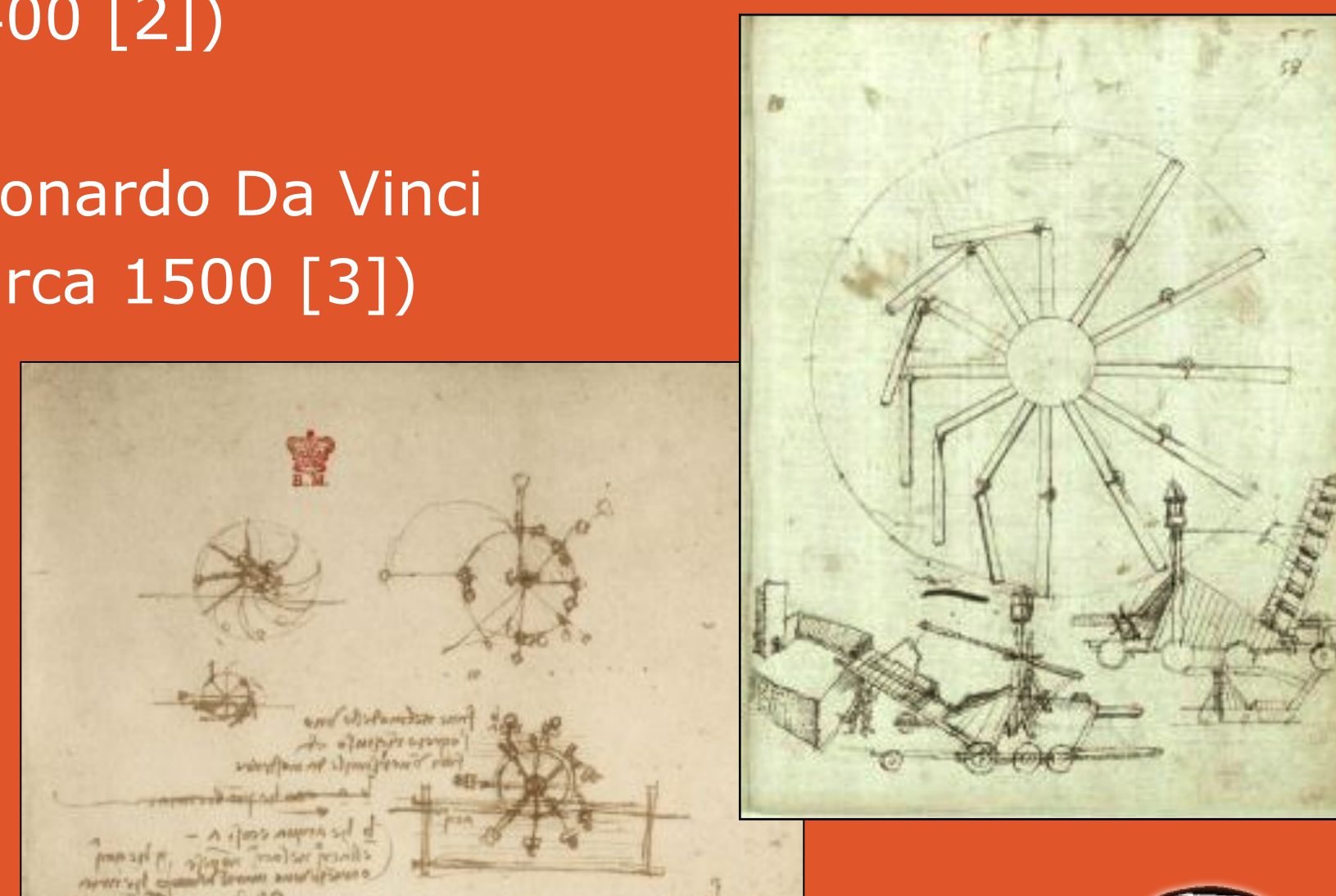


Figure 3: Leonardo Da Vinci sketches of PMMs (Above, Left) Figure 4: Taccola sketch of changing center of mass PMM (Above, Right)

- Robert Boyle (circa 1600 [2])



Figure 5: Drawing of Boyle's "Capillary Bowl" PMM

- This lineage of inquiry became the Study of Thermodynamics (energy transfer) that we know today:

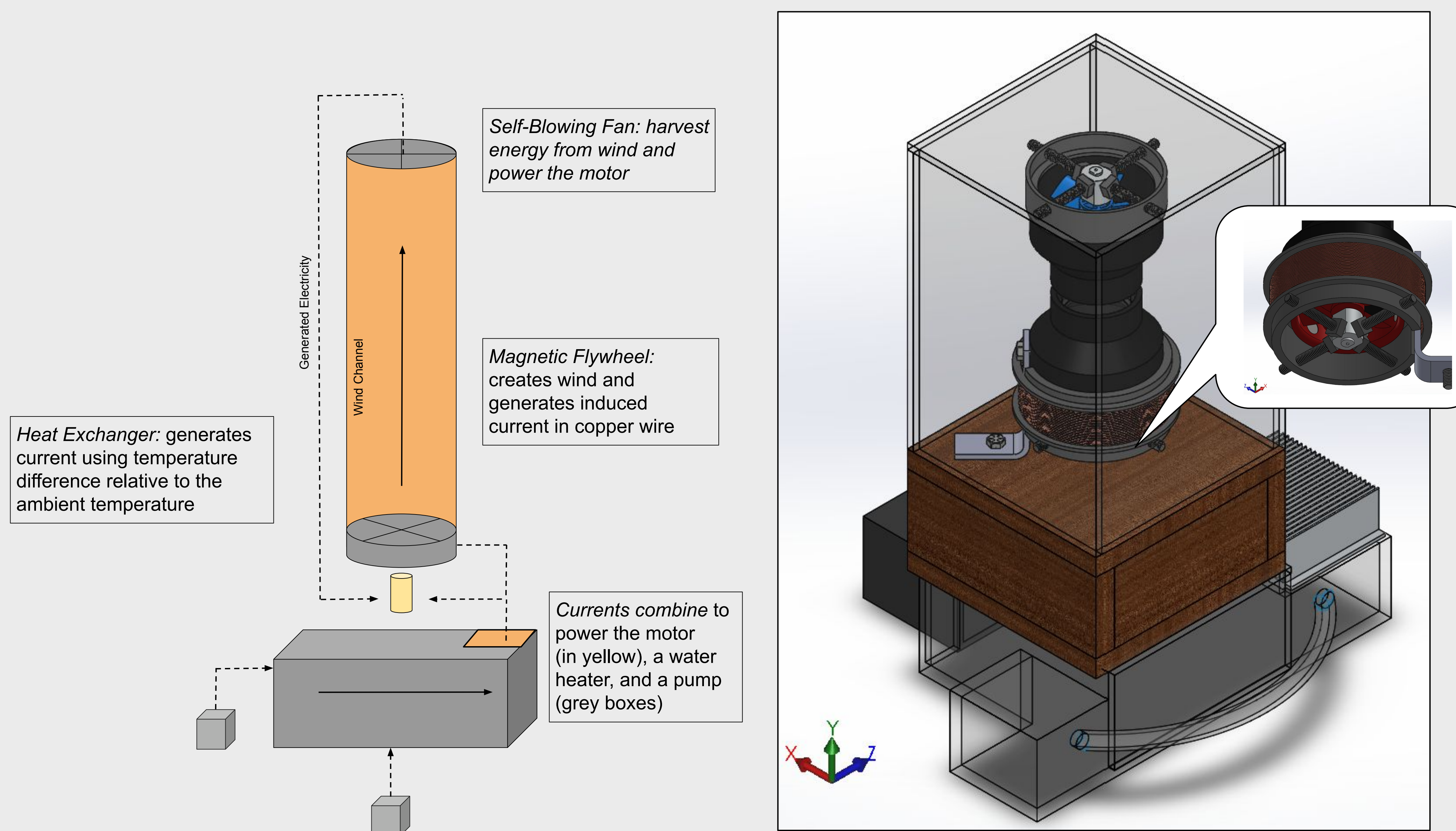
- Consider inefficiencies
 - Excess Heat (Friction, Resistance, etc)
 - Sound
 - Other forms of Vibrations (which do not contribute to the work) are sources of energy loss
- Entropy

MAGIC MOTION: A STUDY IN ENERGY

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Q: What is a Perpetual Motion Machine (PMM)?

A: PMMs are a class of mechanisms which are **powered directly by the energy they create.** With zero energy losses, a PMM will perform work infinitely and... **perpetually** remain in motion!



Poster References

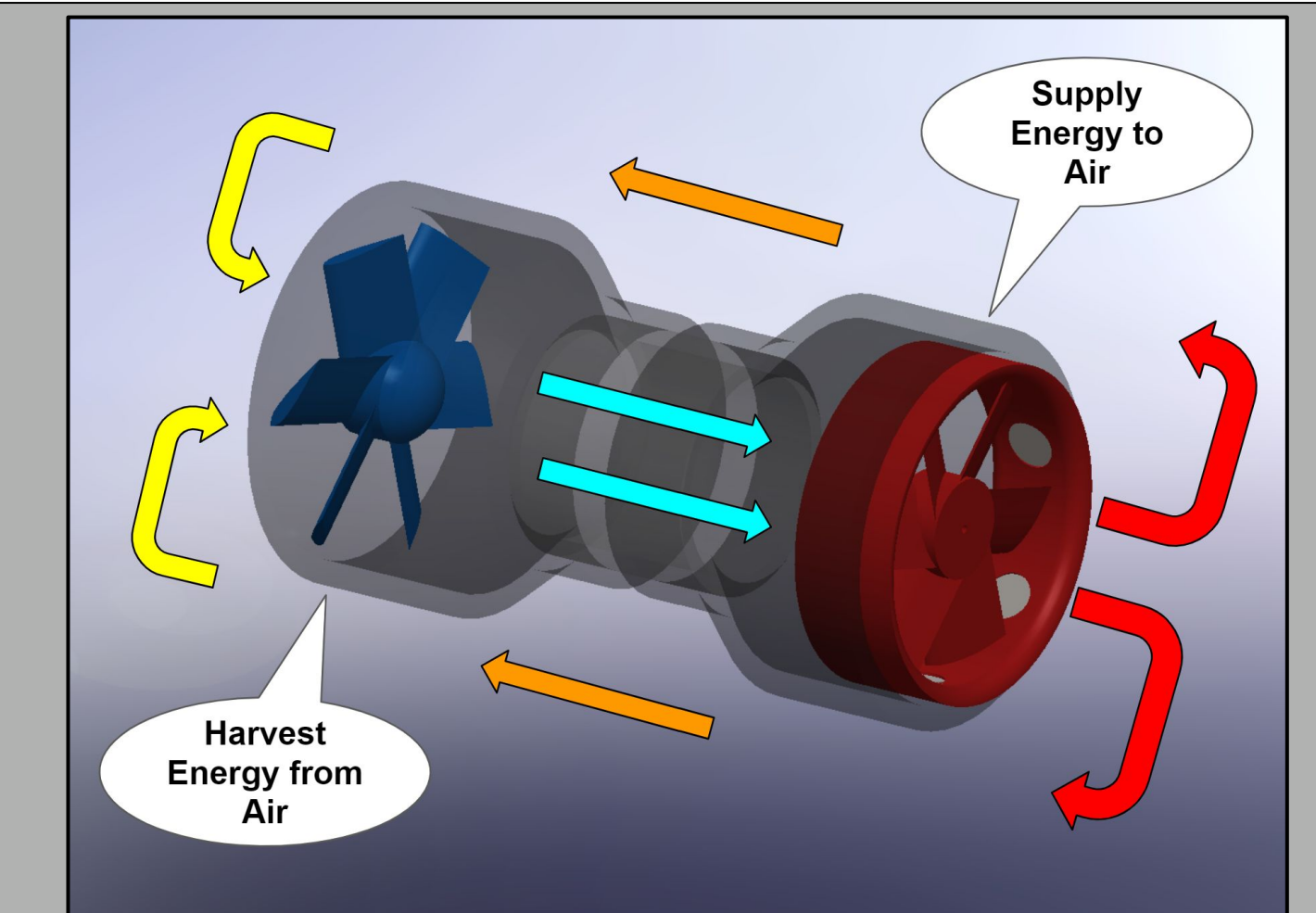
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THEORY + APPLICATION:

- Subsystem I

Self-Blowing Fan

$$Power = \dot{m} \left[\left(\frac{P}{\rho} + \frac{v^2}{2} + gz \right)_{in} - \left(\frac{P}{\rho} + \frac{v^2}{2} + gz \right)_{out} \right]$$



- Subsystem II

Self-Spinning Magnetic Flywheel

- Faraday's Law in electromagnetism

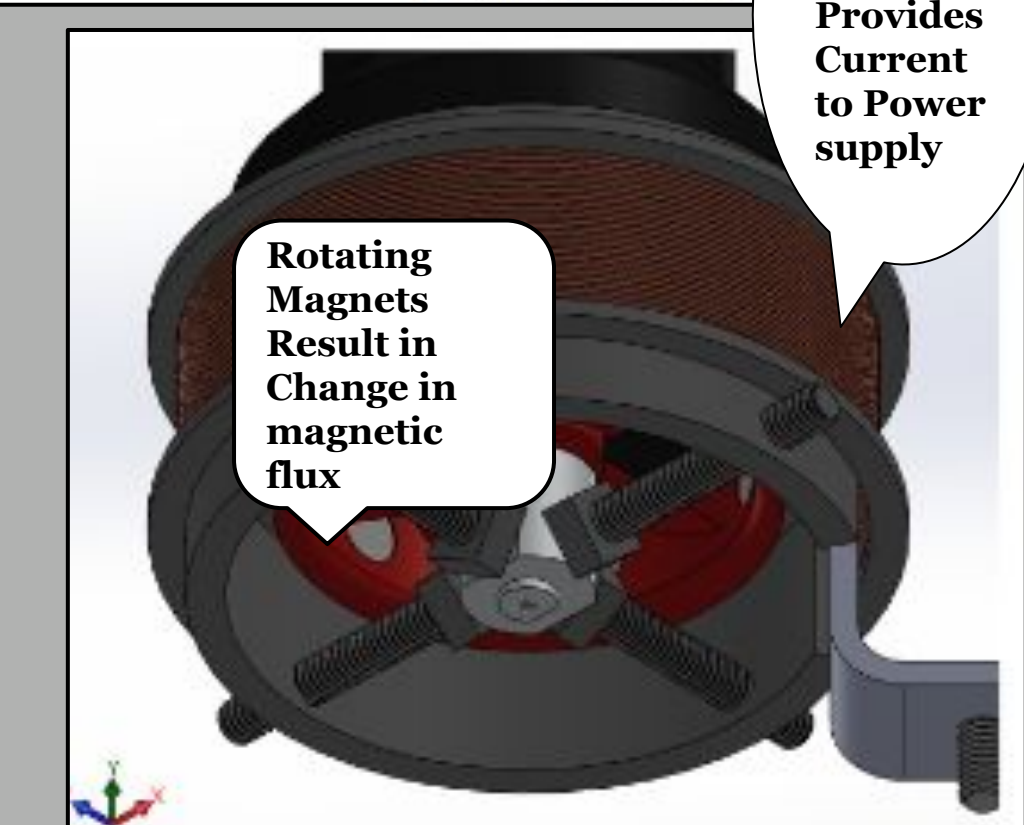
$$\epsilon = -N \frac{\Delta \phi_M}{\Delta t}$$

Faraday's Law

N = number of loops
 $\Delta \phi_M$ = change in magnetic flux
 Δt = change in time
 ϵ = induced voltage (also known as emf)

Relevant Variables:

- \dot{m} = Mass Flow Rate
- C = Specific Heat
- Q = Heat Flow Rate
- T = Temperature
- P = Pressure
- ρ = Density
- v = Velocity
- g = Gravity

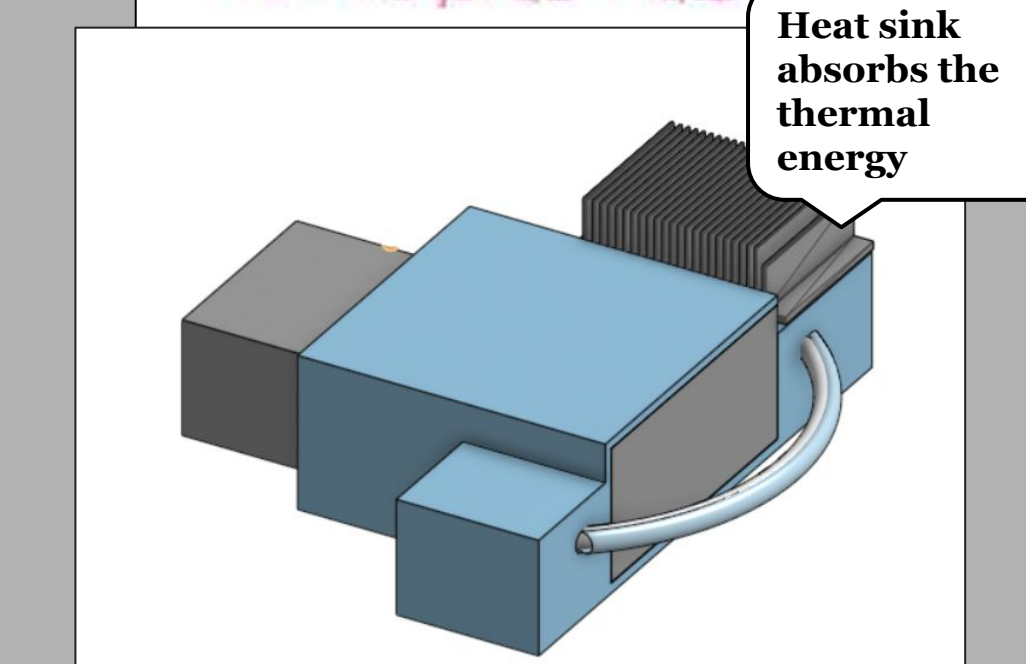


- Subsystem III

Self-Heating/ Self-Pumping Heat Exchanger

$$\dot{Q} = \dot{m}_c C_p (T_{c,out} - T_{c,in})$$

$$\dot{Q} = \dot{m}_h C_p (T_{h,in} - T_{h,out})$$



- By combining all three subsystems, the team studied the role of energy losses within the system as a whole

If you want to learn more, consider:

- What is energy loss?
 - Conservation of Energy
 - Energy cannot be created or destroyed, only transferred or transformed.
 - 2nd Law of Thermodynamics
 - Entropy (chaos, disorder, randomness) in any natural or spontaneous system can only increase or remain constant