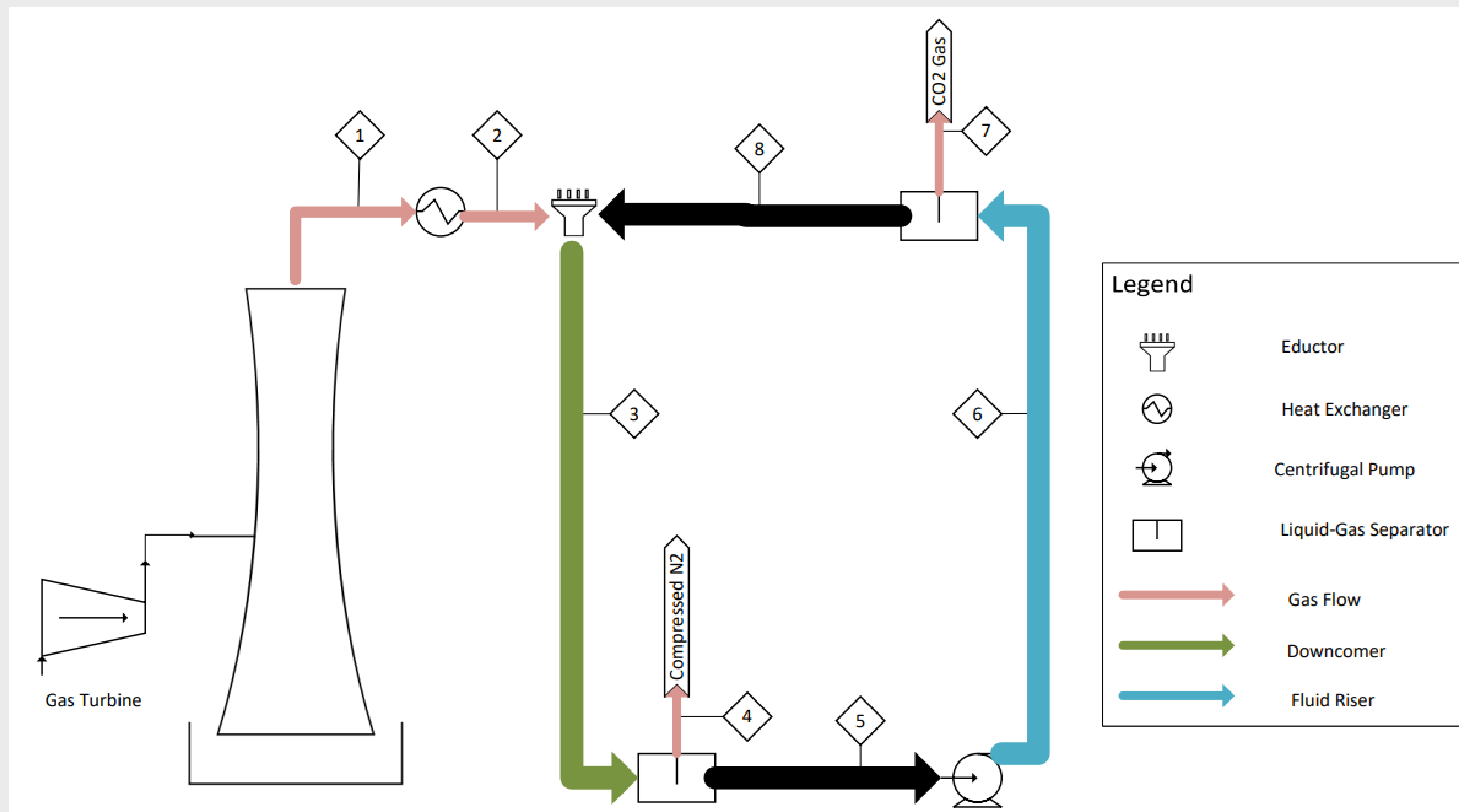


# CO<sub>2</sub> CAPTURE USING HYDRAULIC AIR COMPRESSION

Hydraulic Air Compressor: Isothermal compression and mass transfer of a gas is achieved with a significant height drop of a liquid solvent and gas mixture.



## DOWNCOMER

Temperature-controlled flue gas from the power plant's stack tower is introduced into a mixing point with a high volume stream of water (or other choice of solvent). This forms a two-phase flow of gas entrained in bubbles being dragged down by the solvent. Dropping down a significant height compresses the flow due to hydrostatic pressure, which pressurizes the gas phase.

This is used to separate nitrogen out of the solvent at the bottom of the system, a historic application of the design, while also increasing solubility of CO<sub>2</sub> in the solvent. Both effects combine to capture the greenhouse gas, and prepare it for extraction in the next section.

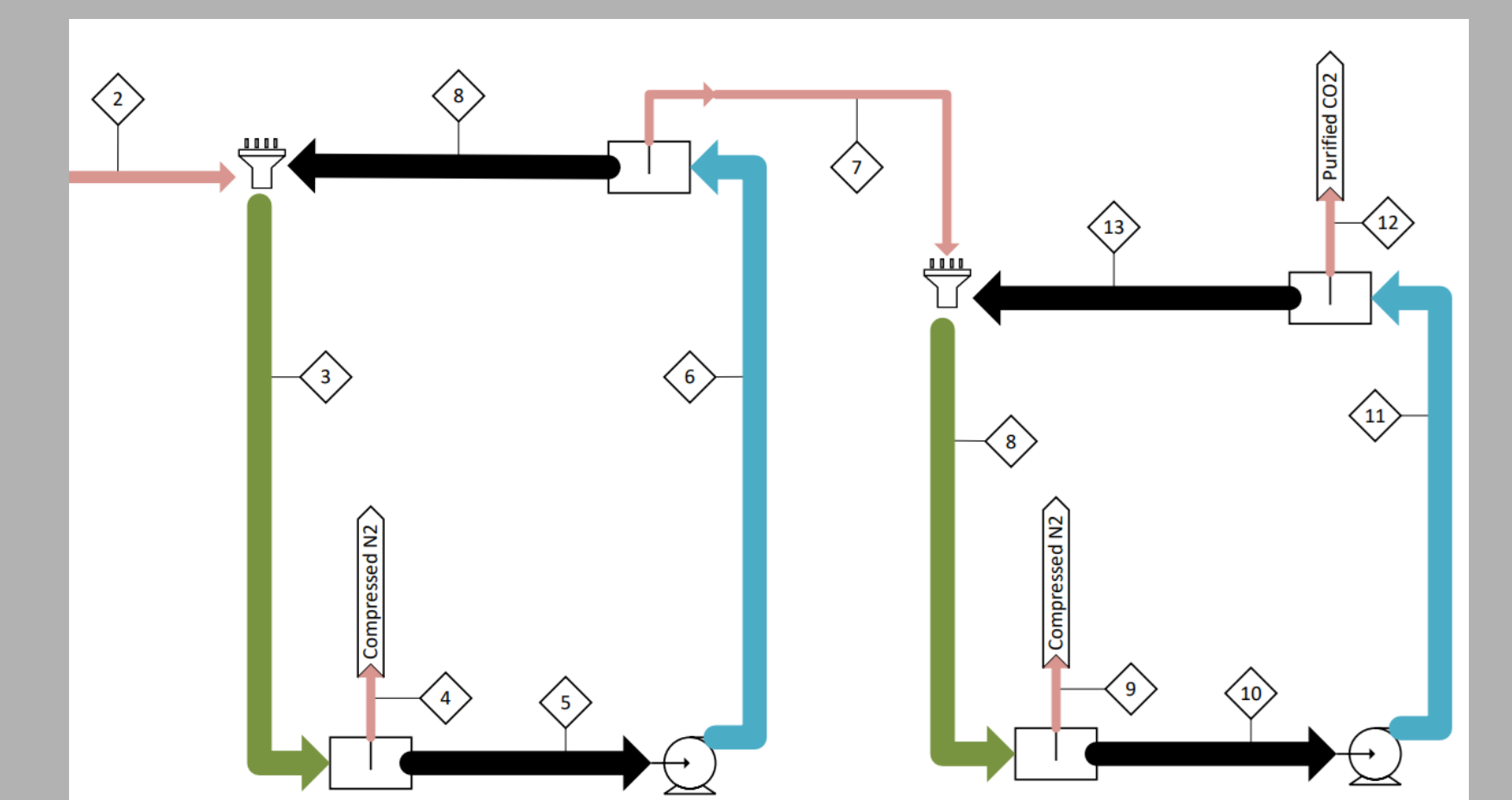
## RISER

The kinetic energy of the water flow will lead it up the riser without the need of outside energy input, most of the way. A pump is included to assist the water flow on the last 6 meters of the flow cycle.

Similar to soda, the dissolved CO<sub>2</sub> will escape the fluid in the riser as atmospheric pressure is reached. A low pressure separator is implemented to capture the gas and direct the fluid back to the mixing point for recycling.

## FUTURE WORK

- Bench-scale experiment: Experimental design will help to solidify scaled efficiency of captured CO<sub>2</sub>.
- Addition of additives: Careful application of additives such as salt or MEA may help to increase solubility of the solvent.
- Locations: The depths of an ocean can be taken advantage of to facilitate this system. Pressure increase down the pipe will be equal with the pressure of ocean depth.
- Multiple stages: Running the extracted CO<sub>2</sub> through additional solvent loops would increase the amount captured from the flue gas.
- Scale up for larger power plants: Design for real-world power plants can provide a measure to reduce carbon emissions.
- Applications of compressed nitrogen: Nitrogen extracted at the high pressure separator is useful as a side-product, or for use as a pressurized coolant gas.
- Other gas sources: Alternative sources of CO<sub>2</sub> or similar gasses can be captured using this system, barring the effects of the gas mixture.



## OBJECTIVES

- Capture carbon dioxide gas that is emitted from a 40 MW power plant without increasing system temperature.
- Reduce maintenance costs using a gravity-based compression design.
- The CO<sub>2</sub> needs to transfer/dissolve into the water using the pressure that is developed as the water mass piles down with simple gravity.
- Separate carbon dioxide from other gases such as nitrogen, argon, oxygen etc...
- Dissolved CO<sub>2</sub> will bubble out whilst flowing up the riser and the pressure decreases to atmospheric. The water flowing up will only need pumping input on the last 15% of the height.

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