# **COLLEGE OF ENGINEERING**



#### Background

- Ammonia used to produce food for half of the global population
- Ammonia manufacturing accounts for 1-3% of the world's energy consumption, 5% of the world's natural gas consumption, and a significant portion of greenhouse gas emissions
- Ammonia is mainly used in the production of fertilizer
- Since it is a toxic and flammable gas, ammonia is very expensive to transport

### **Design Requirements**

- Plant must be able to produce 50 metric tonnes per day of commercial anhydrous ammonia
- Anhydrous ammonia must have a purity of 99.5% by mass
- Ammonia is produced and stored as a liquid at high pressure
- Ammonia is produced from nitrogen and hydrogen intermediates
- Design must employ new modular manufacturing methods
- Minimize the carbon footprint of the plant
- The plant will be located in the Minnesota River Valley

### Acknowledgments

Nick AuYeung Natasha Mallette AIChE - Design Competition committee

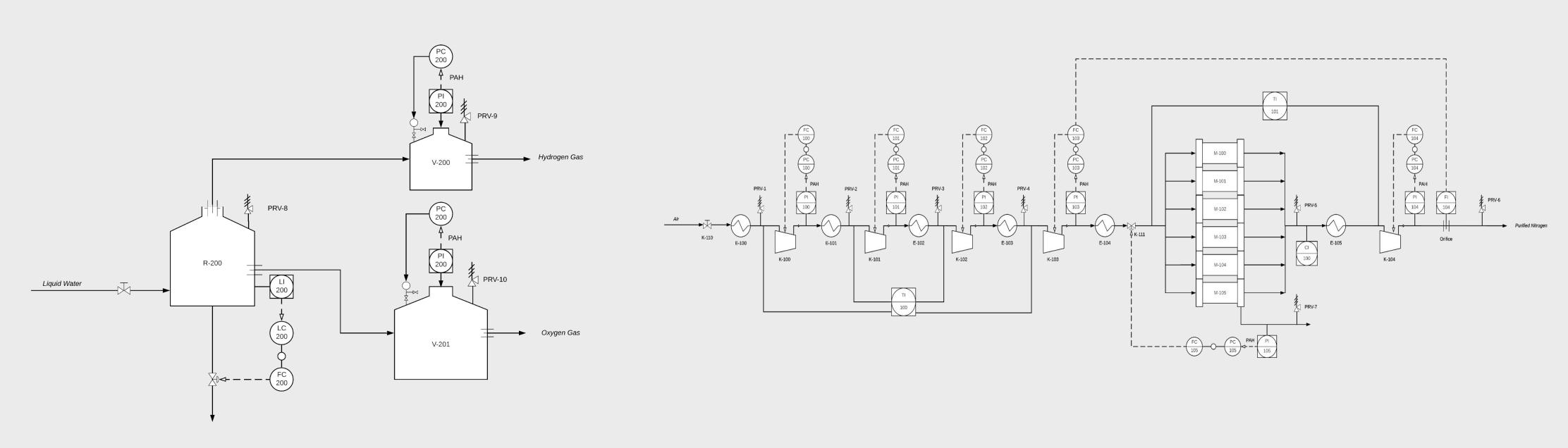
# Chemical, Biological, and Environmental Engineering



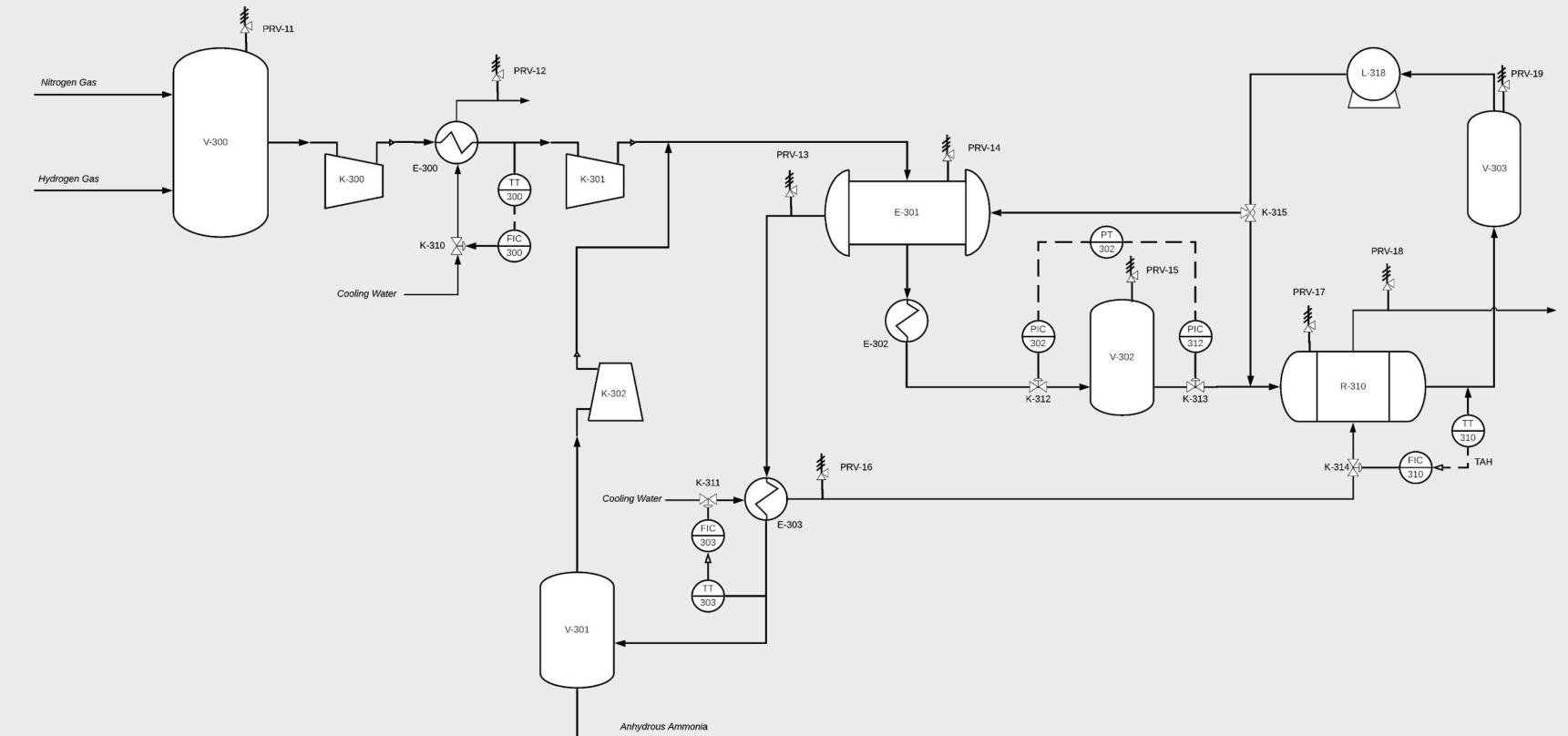
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## **Process & Instrumentation Diagrams**

#### Hydrogen Production



#### Ammonia Production



### **Other Design Considerations**

#### Steam Methane Reformation

•	produces large amounts of $H_2$ gas through	• N
	the water -gas shift reaction	f
•	Inconsistent with the AIChE problem	n
	statement desiring wind powered ammonia	• 5
	production	C
•	The efficiency of the process is reduced with	C
	decreasing size making it incompatible with	• T
	the modular design process	p

Nitrogen Separation

#### Non-thermal Plasma reactors

Non-thermal Plasma reactors create ammonia from room temperature and pressure feed gas making it an inherently safer process Scaling-up from lab scale reactors causes a frop in efficiency because ammonia is decomposed by the plasma This process is prohibitively energy intensive producing 1 gram of ammonia per kilowatt hour

## Safety & Health

- due to smaller loads
- Modular design allows easier transportation
- Smaller quantities produced
- Potentially dangerous chemicals
- Oxygen gas can lead to fires or explosions Ammonia vapor can be toxic
- Hydrogen gas can lead to explosions
- Thorough process safety and proper training is required to ensure the safety of users
- Need to ensure plant is not located near major sources of ground water or rivers
- Potential risk of contaminating waterways • Uses wind power
- Reduced export distance which reduces the use of fossil fuels in transportation

## Economics

- Given ammonia's substantial energy consumption, all energy requirements for the three major units processes (N2 and H2 separation and the ammonia synthesis) will be derived from wind power.
- (1) Single Module Unit 5 MTPD CAPEX: \$13.7 million Ammonia Unit cost: \$5050.26 per MT IRR: 47%
- (2) 10 Modules 50 MTPD CAPEX: \$86.8 million Ammonia Unit Cost: \$4091.43 per MT IRR: 64%
- (3) 10 modules with Turbines: CAPEX: \$108.5 million Ammonia Unit Cost: \$1658.13 per MT IRR: 15%

• Nitrogen gas can cause suffocation

## Environmental Considerations